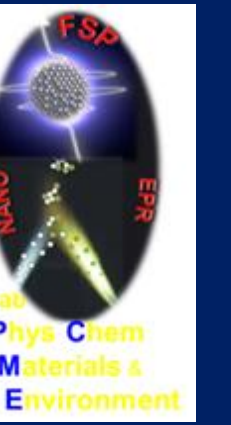


# Titania Quantum Dots Assemblies (Pt/TiO<sub>2</sub>-QDs, Pd/TiO<sub>2</sub>-QDs) by Flame Spray Pyrolysis for Superior Artificial Photosynthesis

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## Introduction

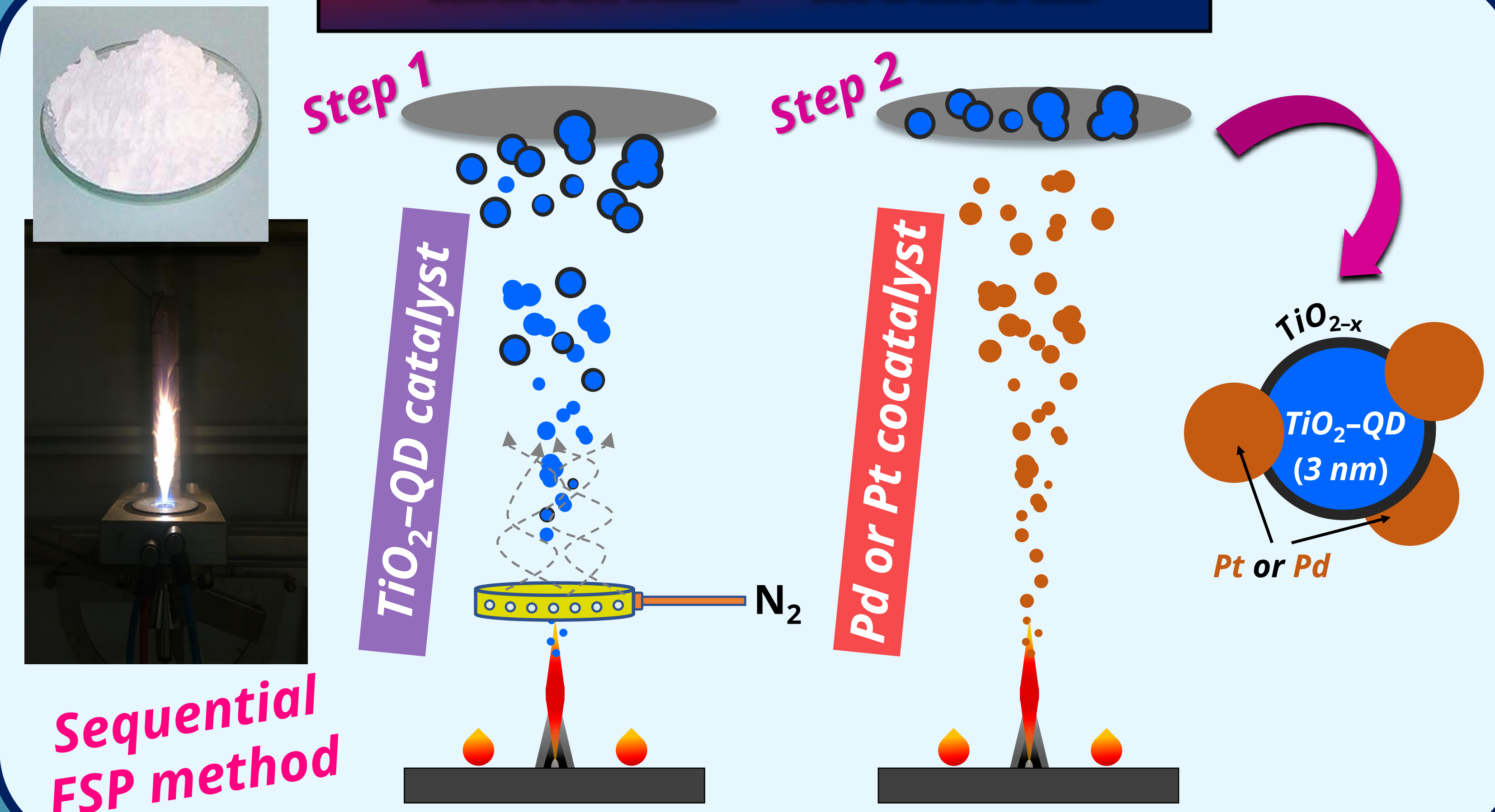
The extraordinary optical and electrical characteristics of quantum dots (QDs), such as their high quantum efficiency, narrow size-dependent absorption, and luminescence windows, have attracted a great deal of attention. Metal oxide quantum dots have a wide range of applications, including photocatalysis (TiO<sub>2</sub>), ultraviolet (UV) protection coatings (ZnO), high-temperature superconductors (CuO) and gas sensing (SnO<sub>2</sub>). The equation that governs the relationship between the crystal size in terms of radius  $R$  and bandgap energy  $E^*$

$$E^* = E_g + \frac{\hbar^2 \pi^2}{2R^2} \left( \frac{1}{m_e} + \frac{1}{m_h} \right) - \frac{1.8e^2}{\epsilon R} - 0.248 E_{\text{Ryd}}$$

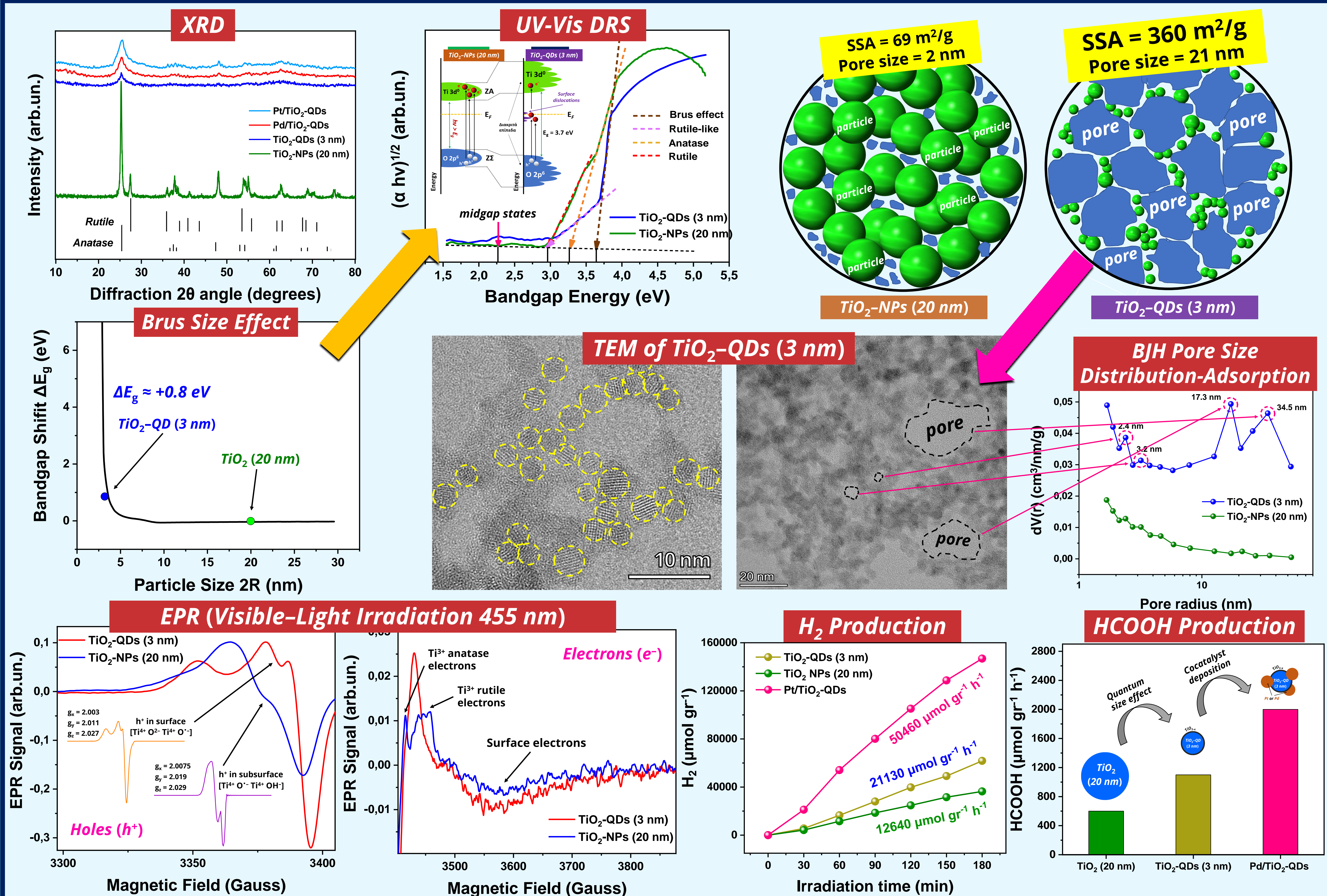
where  $E_g$  is the bulk bandgap energy (at  $R \rightarrow \infty$ ),  $m_e$  and  $m_h$  are the electron and hole effective masses,  $\epsilon$  is the dielectric constant of the material and  $E_{\text{Ryd}}$  the Rydberg energy term. The above equation illustrates that the quantized effect is most significant at sizes below the exciton Bohr diameter but still occurs at bigger sizes.

In the present work, TiO<sub>2</sub> QDs were synthesized and decorated with either 1% Pt or 1% Pd nanoparticles utilizing the innovative sequential Flame Spray Pyrolysis (FSP) method. Finally, these nanoparticles were studied for artificial photosynthesis, i.e. the H<sub>2</sub> production by H<sub>2</sub>O splitting and the HCOOH formation by CO<sub>2</sub> reduction.

## Materials - Methods



## Results - Discussion



## Conclusions

- TiO<sub>2</sub>-QDs of 3 nm were synthesized and decorated with either 1% Pt or 1% Pd using the sequential FSP technology.
- Large pores of approximately 20 and 40 nm were formed; more effective desorption capacity.
- Surface dislocations; under visible light (455 nm), the electrons population was increased due to the quantum size effect on bandgap energy.
- Higher [SSA, porosity, surface electrons population]; the Pt/TiO<sub>2</sub>-QDs and Pd/TiO<sub>2</sub>-QDs display superior H<sub>2</sub> and HCOOH yield.

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